

## Short-term Effects of High CO<sub>2</sub> Concentrations on *Chlorella vulgaris* and *Scenedesmus regularis*

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### ABSTRACT

High CO<sub>2</sub> concentration in the atmosphere can lead to climate change which can pose threats to ecosystems and living organisms. Use of microalgae as biomitigation tool for CO<sub>2</sub> sequestration has become an increasing interests. However, being photoautotrophs, too much of CO<sub>2</sub> can be detrimental to them as well. Thus, the aim of this study is to observe the responses of two microalgae, *Chlorella vulgaris* and *Scenedesmus regularis* after a short-term high CO<sub>2</sub> exposure (i.e. in 484 and 968 ppm CO<sub>2</sub>). The algal maximal quantum yield (i.e. F<sub>v</sub>/F<sub>m</sub>) measured using a chlorophyll fluorometer; content of hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) and level of DNA damage using the Comet assay; were determined. In addition, the medium's pH value was also determined in order to observe the uptake of CO<sub>2</sub> by the algae. Recovery of the algae from the CO<sub>2</sub> stress was also observed. Results showed that supplementing the medium with additional CO<sub>2</sub> induced an acidification reaching a pH between 5.4 and 6.1. After the recovery, the pH increased back into its original alkaline state. The F<sub>v</sub>/F<sub>m</sub> of *C. vulgaris* was reduced after treatments in both [CO<sub>2</sub>] and these conditions did not mitigate after the recovery. Constrastingly, F<sub>v</sub>/F<sub>m</sub> of *S. regularis* was not affected in 484 ppm but showed an increase in 968 ppm CO<sub>2</sub>. Further increase was observed after recovery. The content of H<sub>2</sub>O<sub>2</sub> increased after treatments in *C. vulgaris* but was decreased in *S. regularis* for both [CO<sub>2</sub>]. After recovery showed that the effect was lessen and reduced back to its original content in *C. vulgaris* but started to increase in *S. regularis*. Comet assay analysis showed that most of *C. vulgaris* cells had their DNA damage lies within the highest level of 4 (i.e. highly damaged) after the treatment. In contrast, *S. regularis* cells had their DNA damage at the highest level of 2 (i.e. moderately damaged). Repair activity of the DNA was observed in *C. vulgaris* and *S. regularis* cells treated with 484 ppm but not in 968 ppm CO<sub>2</sub>. *C. vulgaris* was susceptible towards the high CO<sub>2</sub> while *S. regularis* was more tolerant. *S. regularis* has the potential to be effective as biomitigation tools of elevated CO<sub>2</sub>.

**KEYWORDS:** Microalgae, Carbon Dioxide, F<sub>v</sub>/F<sub>m</sub>, H<sub>2</sub>O<sub>2</sub>, DNA Damage

### INTRODUCTION

The problem of climate change arising mainly from carbon dioxide (CO<sub>2</sub>) emission is currently a critical environmental issue. According to that stated in [1], atmospheric CO<sub>2</sub> concentrations have risen from 295 parts per million (ppm) to 380 ppm over the last 100 years. This is mainly contributed by combustion of fossil fuels and from reduced carbon uptake due to global deforestation and loss of arable land. Effects of climate change on the ecosystem are diverse and can be deleterious to living organisms [2, 3]. For example, the emission of high concentration of CO<sub>2</sub> into the atmosphere can cause the shift in carbonate chemistry of seawater or ocean acidification. Ocean acidification affects, among others, the feeding rates in a scleractinian coral [4], reproduction of coral reef fishes [5] and immune responses of oysters [6]. Photosynthetic organisms rely on CO<sub>2</sub> to grow and develop, thus, they are also deemed to be highly affected by the high concentrations of atmospheric CO<sub>2</sub>. Growth of some C<sub>3</sub> plants such as *Caragana microphylla*, a leguminous shrub and *Stipagrandis*, a grass

was stimulated by elevated CO<sub>2</sub> but the biomass of a C<sub>4</sub> grass, *Cleistogenes squarrosa* was not affected by this [7]. Nitrate assimilation was also observed to be slower in wheat grown under elevated CO<sub>2</sub> [8]. Furthermore, elevated CO<sub>2</sub> enhanced the production of chlorophylls, proteins, phenolic compounds as well as carotenoids and alkaloids [9].

There is a growing interest in using seaweeds and microalgae as a tool for green technology in mitigating the effect of elevated CO<sub>2</sub> concentrations. Being photoautotrophs, they utilize CO<sub>2</sub> for photosynthesis which can be useful in CO<sub>2</sub> sequestration [1,10]. In addition, their capacity in fixing CO<sub>2</sub> is much greater than C<sub>4</sub> plants [10]. CO<sub>2</sub> fixation by the algae can also be combined with production of biomass which can be burned for heat and electricity [11, 12]. They can be good bioindicators as well since they are sedentary, easy to identify, have rapid reproduction rates, very short life span and are widely distributed in the aquatic ecosystem [13]. Moreover, algae are known to be able to adapt to harsh environments by inducing protective mechanisms which can be beneficial as bioindicator parameters [14]. This paper describes the biochemical and physiological responses of two microalgae, *Chlorella vulgaris* and *Scenedesmus regularis* against the effect of elevated CO<sub>2</sub> concentration under the laboratory conditions. From the observations, potential candidate for efficient CO<sub>2</sub> mitigation among the two algae was also determined.

## MATERIALS AND METHODS

### Algal Materials and CO<sub>2</sub> Treatments:

The green microalgae, *Chlorella vulgaris* (UMT-M1) and *Scenedesmusregularis*(KS-MC1) were cultured in f/2 media [15] and cultivated for treatment at log phase of growth. The algae in filtered seawater were aerated with two different concentrations of CO<sub>2</sub> (i.e. 484 and 968 ppm) for 6h at 25±1°C under a white fluorescent light. After the treatment, the CO<sub>2</sub> aeration was stopped and the algae were aerated with normal air and left at room temperature for a 24h recovery period. Control was subjected to similar conditions as treatment but with no additional CO<sub>2</sub>.

### Parameters Analyzed:

Chlorophyll (chl) a fluorescence was measured in terms of the dark-adapted maximal quantum yield (i.e.  $F_v/F_m$ ) with a handheld chlfluorometer, AquaPen-P AP-P 100 (Photon Systems Instruments, Czech Republic). At the start of the measurement, a short, red, actinic pulse (~3000  $\mu\text{mol m}^{-2} \text{s}^{-1}$  at 655 nm) was prompted for 5 s to ensure a stabilized fluorescence emission during the following  $F_m$  measurement. Then  $F_0$  was measured with a pulsed, blue measuring light (~900  $\mu\text{mol m}^{-2} \text{s}^{-1}$ , 455 nm), and  $F_m$  was determined with a saturating white light pulse (~3000  $\mu\text{mol m}^{-2} \text{s}^{-1}$ ). The maximal quantum yield,  $F_v/F_m$  was calculated as  $(F_m - F_0)/F_m$ .

The H<sub>2</sub>O<sub>2</sub> content in the algae was determined according to [16]. The treated cells were first harvested by centrifugation at 10,000 rpm for 10 min at 4°C. The cells were washed once with distilled water and recentrifuged. The cell pellet was subjected to wet weight estimation and then dried in the oven for 2 h at 80°C. Algal material was homogenized with 10% (w/v) trichloroacetic acid in an ice bath. Then, the homogenate was centrifuged at 7,000 g for 10 min, and the supernatant was added with 50 mmol/L potassium phosphate buffer (pH 7.0) and 1mol/L potassium iodide. The absorbance of the mixture was measured at 390 nm. A series of 30% H<sub>2</sub>O<sub>2</sub> concentrations was used to construct a standard curve and the algal H<sub>2</sub>O<sub>2</sub> content was determined from this.

Alkaline comet assay was performed according to [17]. After centrifugation at 2000 rpm for 5 min, cell pellets were embedded in a layer of 0.5% low-melting agarose on a fully frosted microscopic slide coated with 0.5% normal-melting agarose (NMA). The slide was placed in a lysis solution containing 300 mM NaOH, 30 mM Na<sub>2</sub>EDTA and 0.01% SDS for 5 min and then in an electrophoresis buffer (300mM NaOH plus 1mM Na<sub>2</sub>EDTA) for 20 min at 4°C to allow unwinding of DNA. After that, electrophoresis was conducted with the same buffer at 4°C for 20 min at 25V and 300 mA. The slides then were neutralized by immersing twice in a neutralizing buffer (400 mM Tris pH 7.4) for 5 min. The DNA molecules then were stained with ethidium bromide for 5 mins and dipped in chilled distilled water to remove excess stain. The migrated DNA or comet images were observed under an epi-fluorescent microscope. Tail length and tail moment were determined using an image analysis software as an indication of DNA damage. Data was taken from 50 cells per sample. Scores of DNA damage were determined according to that categorized by [18]: score 0 (no damage, <1% DNA in tail), score 1 (low damage, 1-25% DNA in tail), score 2 (medium damage, >25-45% DNA in tail), score 3 (high damage, >45-70% DNA in tail) and score 4 (very high damage, >70% DNA in tail).

### Statistical Analysis:

Values of all the parameters tested were related to 100% of controls for better comparison unless otherwise stated. Mean values and standard error were determined from three replicates of each treatment. The statistically significant differences between the CO<sub>2</sub> concentrations within similar algal species were analyzed using a Student's t-Test and one-way ANOVA followed by Tukey HSD *post-hoc* test at probability level of 0.05.

## RESULTS AND DISCUSSION

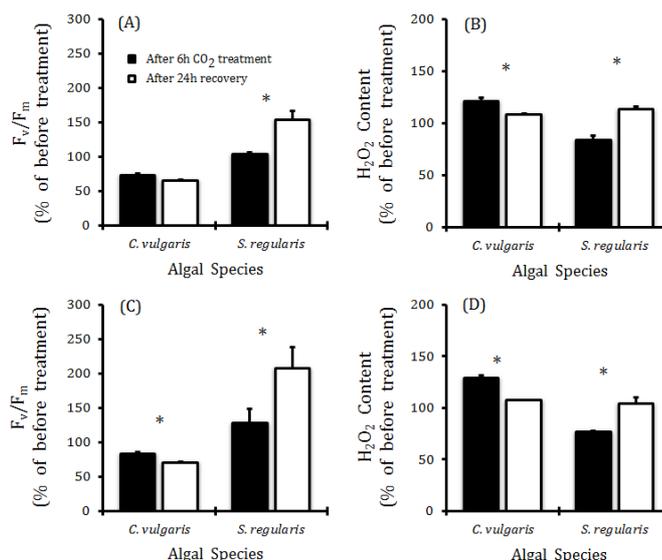
CO<sub>2</sub> when diluted in aqueous solution forms carbonic acid (H<sub>2</sub>CO<sub>3</sub>). This H<sub>2</sub>CO<sub>3</sub> dissociates in the water releasing hydrogen ions (H<sup>+</sup>) and bicarbonate (HCO<sub>3</sub><sup>-</sup>), thus, increasing acidity or a decrease in pH of the medium as shown by both algae after the 6h CO<sub>2</sub> treatment (Table 1). This condition creates acidification in the medium and can result in the decrease efficiency of bicarbonate uptake by the algae [19]. In response to low CO<sub>2</sub> in the cells, microalgae activate an active transport system known as the CO<sub>2</sub>-concentrating mechanism (CCM) to enhance their photosynthetic performance in inadequate CO<sub>2</sub> supply [20]. However, it is expected that increases in dissolved CO<sub>2</sub> cause down-regulation of microalgal CCM capacity [21, 22]. Thus, this will generate stress in the cells. Being photoautotrophs, *C. vulgaris* and *S. regularis* use the CO<sub>2</sub> available in the medium for photosynthesis and photosynthesis performed in water will always cause an increase in pH since CO<sub>2</sub> in the water is withdrawn in the form of H<sup>+</sup> associated with HCO<sub>3</sub><sup>-</sup>. Hence, an increase in pH was observed after the 24h recovery period (Table 1). This change in pH indicate that both algae were actively taking in the CO<sub>2</sub> in the medium during the recovery period.

**Table 1:** pH of medium after the 6h CO<sub>2</sub> treatment and after 24h recovery as compared with control. Data are mean±S.E. Different letters after value indicate statistically significant differences at p<0.05 (ANOVA, Tukey HSD *post-hoc* test, n=3).

Treatment Conditions	Algal Species and CO <sub>2</sub> Concentrations			
	<i>C. vulgaris</i>		<i>S. regularis</i>	
	484 ppm	968 ppm	484 ppm	968 ppm
Control	8.7 ± 0.04 <sup>A</sup>	8.5 ± 0.03 <sup>B</sup>	7.6 ± 0.03 <sup>B</sup>	7.7 ± 0.01 <sup>B</sup>
After 6h CO <sub>2</sub>	6.0 ± 0.03 <sup>B</sup>	6.1 ± 0.02 <sup>C</sup>	5.6 ± 0.01 <sup>C</sup>	5.4 ± 0.01 <sup>C</sup>
After 24h recovery	8.8 ± 0.04 <sup>A</sup>	8.7 ± 0.04 <sup>A</sup>	7.8 ± 0.02 <sup>A</sup>	7.9 ± 0.05 <sup>A</sup>

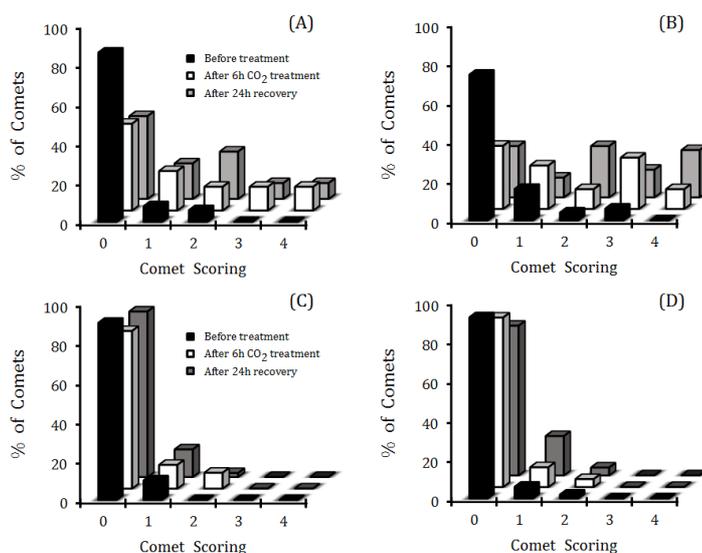
F<sub>v</sub>/F<sub>m</sub> of *C. vulgaris* after the CO<sub>2</sub> treatment showed a significant reduction of about 27% and 17% from that of untreated algae in 484 (p=0.015) (Fig. 1a) and 968 (p=0.044) (Fig. 1c) ppm CO<sub>2</sub>, respectively. Contrastingly, F<sub>v</sub>/F<sub>m</sub> of *S. regularis* increased significantly compared to untreated algae with 28% increment in 968 ppm (p=0.013) (Fig. 1c) but remained on par with that of untreated alga in 484 ppm CO<sub>2</sub> (Fig. 1a). At the physiological level, the measurement of F<sub>v</sub>/F<sub>m</sub> is an effective parameter to monitor the efficiency of light utilization by the algae as well as to assess the efficiency and stability of the photosystem II (PSII), a major component of the photosynthetic machinery [23]. F<sub>v</sub>/F<sub>m</sub> values around 0.6–0.7 are expected for healthy microalgal cells, whereas lower values are observed if cells have been exposed to biotic or abiotic stress [24]. Therefore, it can be concluded that *C. vulgaris* cells were stressed under the high [CO<sub>2</sub>] but not *S. regularis* and high [CO<sub>2</sub>] can damaged the PSII of *C. vulgaris* resulting in inhibition of the transfer of electrons, leading to side effects such as oxidative stress [25]. In contrast, an increase in F<sub>v</sub>/F<sub>m</sub> as shown by *S. regularis* under both [CO<sub>2</sub>] can indicate increase in the efficiency of PSII and also reduced risk of damage caused to PSII by oxidative stress. Similar increase in F<sub>v</sub>/F<sub>m</sub> was also reported by [26] with soybean under 1000 ppm CO<sub>2</sub>, [27] with white birch under 700 ppm CO<sub>2</sub> and [9] with periwinkle under 560 ppm CO<sub>2</sub>. The F<sub>v</sub>/F<sub>m</sub> did not increased back to normal after the 24h recovery period in *C. vulgaris* treated with both [CO<sub>2</sub>] (Fig. 1a,c) indicating that the alga experienced chronic photoinhibition whereby the rate of PSII damage is faster than the rate of repairment [28]. In contrast, a further increment in F<sub>v</sub>/F<sub>m</sub> was observed significantly in *S. regularis* treated with both [CO<sub>2</sub>] after the recovery period (Fig. 1a,c).

The content of H<sub>2</sub>O<sub>2</sub> was observed to be significantly increased after the CO<sub>2</sub> treatment and decreased back to normal values after the 24h recovery treatment (Fig. 1b,d). Under stress conditions, the photosynthetic carbon fixation and electron transport rates are not of similar magnitude. This can create an electron excess that cannot be consumed, which will react with O<sub>2</sub>, producing reactive oxygen species (ROS), including H<sub>2</sub>O<sub>2</sub> [29]. Damaged PSII as indicated by lower F<sub>v</sub>/F<sub>m</sub> in *C. vulgaris* can also create ROS [25]. Thus, it is expected to observe an increase in H<sub>2</sub>O<sub>2</sub> in this alga. It is reported that biological toxicity of H<sub>2</sub>O<sub>2</sub> appears to be through oxidation of thiol (-SH) groups and can be enhanced [30]. Following oxidative conditions, algae, like many other living organisms respond by increasing antioxidant defences (e.g. [31, 32]). These antioxidants scavenge and neutralize the ROS lowering the effect of oxidative stress. Hence, the back to normal value observed in *C. vulgaris* after the recovery period (Fig. 1b,d). Contrastingly, a 17% and 24% reduction in H<sub>2</sub>O<sub>2</sub> content was observed in *S. regularis* treated with 484 (Fig. 1b) and 968 (Fig. 1d) ppm CO<sub>2</sub>, respectively. In addition to the results of F<sub>v</sub>/F<sub>m</sub> (Fig. 1a,c) and cell viability (data not shown), it can be assumed that the extra CO<sub>2</sub> was not a rate-limiting to the photosynthetic processes of *S. regularis* and the alga uses this CO<sub>2</sub> to increase its ability in maintaining life. Furthermore, at low concentration, H<sub>2</sub>O<sub>2</sub> can act as a signal molecule involved in the regulation of specific biological processes and triggering tolerance against various environmental stresses [33].



**Fig. 1:** Maximal quantum yield and  $H_2O_2$  content of *C. vulgaris* and *S. regularis* after 6h treatment with 484 (A-B) and 968 (C-D) ppm  $CO_2$  and after 24h recovery. Statistical analysis was done by comparing between treatment and recovery periods within similar species (Student's t-Test,  $n=3$ ,  $p < 0.05$ ).

ROS are known to have damaging effects on many important biomolecules including DNA (as reviewed by [34]). In fact,  $H_2O_2$ , itself has been proven to induce DNA single and double-strand breaks in cells [35]. From the comet assay (Fig. 2), it is evident that high  $[CO_2]$  induced DNA damage in both microalgae. Following exposure to high  $[CO_2]$ , the number of comets belonging to scores of 3 and 4 increased significantly in *C. vulgaris* (Fig. 2a,b). It was observed that high  $[CO_2]$  induced the production of  $H_2O_2$  in this alga as well (Fig. 1b,d). Thus, it can be said that the damage of DNA at scores 3 and 4 may be because of the  $H_2O_2$  accumulated in the cells of *C. vulgaris*. After the recovery, the cells that belong to scores 3 and 4 decreased 8% in 484 ppm (Fig. 2a) compared to after the treatment. The reduction may indicate that the cells have repaired the  $H_2O_2$ -induced DNA single-strand breaks during the recovery or may also indicate that the highly damaged cells have gone into apoptosis and that the fragmented DNA has disappeared from the gel, leaving only relatively undamaged cells [36]. This condition, however, was not observed for cells treated with 968 ppm  $CO_2$  since the % of cells belonging to scores 3 and 4 increased 2% after the recovery (Fig. 2b). In comparison, *S. regularis* treated with 484 (Fig. 2c) and 968 (Fig. 2d) ppm  $CO_2$  showed significantly less DNA damage with the majority of the comets falling into scores 0, 1 and 2. DNA repair activity was also observed during recovery for *S. regularis* cells treated with 484 ppm  $CO_2$  whereby there was a 4% increase and a 6% decrease in cells belonging to score 0 and 2, respectively as compared to after treatment (Fig. 2c).



**Fig. 2:** Levels of DNA damaged in *C. vulgaris* cells after 6h treatment with 484 (A) and 968 (B) ppm  $CO_2$  (white bars) and after 24h recovery (grey bars) compared to control (black bars) evaluated using Comet assay.

### Conclusion:

In conclusion, *C. vulgaris* was stressed by the high [CO<sub>2</sub>] as the CO<sub>2</sub> becomes toxic to the alga, inducing generation of H<sub>2</sub>O<sub>2</sub> which in turn affecting the photosynthetic PSII apparatus and damaging the DNA. It also appeared that *C. vulgaris* was not able to fully recover even though the stress subsided. On the other hand, *S. regularis* responded differently from *C. vulgaris*. *S. regularis* was not stressed by the presence of the 'extra' CO<sub>2</sub> and seemed to be growing healthily in these conditions. Therefore, it can also be concluded that while *S. regularis* is a high CO<sub>2</sub>-tolerant species, *C. vulgaris* is more susceptible towards the CO<sub>2</sub>. Additionally, *S. regularis* can be an efficient candidate as biomitigation tool of elevated CO<sub>2</sub>. Furthermore, the findings from this study can also be used as bioindicators which can help environmentalists to keep track of the current increase in atmospheric CO<sub>2</sub>.

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